- 1 Immobilisation and characterisation of biocatalytic co-factor recycling
- 2 enzymes, glucose dehydrogenase and NADH oxidase, on aldehyde functional
- 3 ReSyn[™] polymer microspheres

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Abstract

- The use of enzymes in industrial applications is limited by their instability, cost and
- 21 difficulty in their recovery and re-use. Immobilisation is a technique which has been
- 22 shown to alleviate these limitations in biocatalysis. Here we describe the
- 23 immobilisation of two biocatalytically relevant co-factor recycling enzymes, glucose
- 24 dehydrogenase (GDH) and NADH oxidase (NOD) on aldehyde functional ReSyn™
- 25 polymer microspheres with varying functional group densities. The successful
- 26 immobilisation of the enzymes on this new high capacity microsphere technology
- 27 resulted in the maintenance of activity of ~40% for GDH and a maximum of 15.4%
- for NOD. The microsphere variant with highest functional group density of ~3 500

µmol.g⁻¹ displayed the highest specific activity for the immobilisation of both enzymes at 33.22 U.mg⁻¹ and 6.75 U.mg⁻¹ for GDH and NOD with respective loading capacities of 51% (0.51 mg.mg⁻¹) and 129% (1.29 mg.mg⁻¹). The immobilised GDH further displayed improved activity in the acidic pH range. Both enzymes displayed improved pH and thermal stability with the most pronounced thermal stability for GDH displayed on ReSyn™ A during temperature incubation at 65 °C with a 13.59 fold increase, and NOD with a 2.25-fold improvement at 45 °C on the same microsphere variant. An important finding is the suitability of the microspheres for stabilisation of the multimeric protein GDH.

1. Introduction

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- Biocatalysis is gaining momentum due to the specificity, regio- and enantioselectivity of enzymes over chemical synthetic routes, and are finding application in the production of fine chemicals and pharmaceuticals [1, 2]. However, the realisation of these applications is still limited by their relative instability in desirable reaction systems, including extremes of pH and elevated temperatures [3-7]. A further limitation is their current cost, compounded by difficulties in recovery and re-use of the biocatalyst [8, 9].
- 47 Immobilisation of the biocatalyst has to an extent alleviated these limitations by 48 reducing the complexity of enzyme recovery and facilitating re-use of the enzyme [5, 49 6, 8, 9]. Immobilisation frequently results in the enhancement of thermal and pH 50 stability, and has previously resulted in improved specificity of the biocatalyst [5, 7-51 12]. A problem associated with solid supports is their relatively low loading capacity 52 and hence low volumetric activity (high non-catalytic load), resulting in the dilution of 53 specific and volumetric activity of the catalyst [13, 14]. Support based systems that 54 display a high loading capacity is desirable since these can overcome identified 55 limitations of high non-catalytic load such as [15]. The various methods and 56 techniques used to immobilise enzymes is the subject of reviews by Brady and 57 Jordaan [14]; Cao et al. [8]; Sheldon [7]; Spahn and Minteer [16]; Tischer and 58 Kasche [17].
 - Oxidoreductases (E.C. 1) are an important class of biocatalyst, and make up about one quarter of all known enzymes [9]. In addition to the various generic limitations of enzymes in large-scale applications, this enzyme classification further requires stoichiometric quantities of expensive nicotinamide co-factors [9, 18]. To reduce the cost associated with the co-factor, this co-substrate is required to be re-used and recycled during biocatalytic reactions [9, 19]. This is suitably achieved through the use of enzymes such as glucose dehydrogenase (GDH, E.C. 1.1.1.47; Fig. 1(a)) and NADH oxidase (E.C. 1.6.3.1; Fig. 2(b)) for recycling these nicotinamide co-factors [20-24].
- There are few reported cases in the literature relating to the immobilisation of these enzymes, particularly GDH. This may be owing to the proposed difficulties

associated with the immobilisation of multimeric enzymes, such as distortion or denaturing, leading to loss in activity [25, 26]. This may be solved through the use of supports which offer multipoint attachment [5]. In this work, we evaluate the immobilisation and characterisation of the biocatalytic co-factor recycling enzymes GDH and NOD, on aldehyde functional ReSyn™ polymer microspheres of varying functional group density. We further characterise the stability of the immobilised enzymes in comparison to the native enzymes in varying pH and temperature conditions.

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2. Materials and Methods

2.1. Chemicals and Reagents

Bovine serum albumin (BSA; fraction V; 98%) was purchased from Carl Roth, βnicotinamide adenine dinucleotide (NADH), flavin adenine dinucleotide disodium salt hydrate (FAD), potassium dihydrogen orthophosphate (KH₂PO₄), di-potassium hydrogen orthophosphate (K₂HPO₄), sodium chloride (NaCl), hydrochloric acid (HCl: 38%), D(+)-glucose, boric acid, citric acid, di-sodium hydrogen orthophosphate (Na₂HPO₄), potassium hydroxide (KOH) and NADH oxidase (NOD - Bacillus licheniformis) were purchased from Sigma-Aldrich. Glucose dehydrogenase (GDH 102 - Bacillus megaterium) was purchased from Codexis (USA). β-Nicotinamide adenine dinucleotide free acid Grade I (NAD+) was obtained from Roche. Quick start Bradford protein assay reagent (including bovine gamma globulin protein standard) was purchased from Bio-Rad Laboratories. Three variants of aldehyde functional ReSyn™ microspheres were gifted by ReSyn™ Biosciences (South Africa). The polymer microspheres varied in their functional group densities of 3500 µmol.g⁻¹, 1200 µmol.g⁻¹ and 300 µmol.g⁻¹. The particle size distributions for these microspheres varied from 10.88 \pm 3.37 μ m, 7.12 \pm 0.62 μ m and 6.65 \pm 1.36 μ m respectively. For the purposes of this study these variants were named A, B and C respectively. The microspheres were supplied in 5 ml aqueous suspensions.

2.2. Enzyme Preparation

GDH (2.5 ml of 5 mg.ml⁻¹) was desalted using PD-10 gel filtration columns (GE Healthcare) according to the manufacturers' protocol to remove preservatives which could potentially interfere with immobilisation. The protein concentration was assayed using Bradford Quick Start protein assay reagent (Bio-Rad) using Bovine Gamma Globulin (BGG) as the standard protein. The GDH enzyme solution was diluted to a final concentration of 1 mg.ml⁻¹. NADH oxidase was supplied as a pure protein extract and was therefore not de-salted before immobilisation.

2.3. Protein Quantification

Protein was quantified by the method of Bradford [27] using Quick Start Bradford assay reagent (Bio-Rad). Standard curves were generated using BSA as the standard. Briefly, dye reagent, 250 μ l, was added to 5 μ l of sample and incubated for 5 min. The assay was quantified spectrophotometrically at 595nm. All assays were performed in triplicate and data are presented as the mean \pm standard deviation.

2.4. Enzyme Assays

2.4.1. Glucose Dehydrogenase Enzyme Assay

Activity for both native and immobilised GDH was determined by following the kinetic reduction of NAD $^+$ to NADH using spectrophotometric absorbance at 340 nm and monitored on a microtitre plate spectrophotometer (Biotek, PowerWave HT). The assay reagent contained 1 mM NAD $^+$, 50 mM D(+)-glucose and 50 mM Tris-HCl buffer at pH 8.0. One Unit of GDH was defined as the amount of enzyme required to reduce 1 µmol of NAD $^+$ per minute at 37 °C. The enzyme reactions contained 5 µl of GDH solution with 195 µl of assay reagent. Microspheres with bound enzyme were washed twice with 1 ml of MilliQ H $_2$ O and resuspended to 100 µl in 20 mM Tris-HCl pH 8.0. Particle preparations were diluted to within the dynamic range of the kinetic assay (as determined using commercial GDH).

2.4.2. NADH Oxidase Enzyme Assay

Activity for both native and immobilised NOD was determined by following the kinetic oxidation of NADH to NAD $^+$ via the same spectrophotometric absorbance used in section 2.4.1. The oxidation of NADH to NAD $^+$ requires the concomitant conversion of oxygen to hydrogen peroxide/water. Assay reagent contained 1 mM NADH, 0.1 mM FAD and 50 mM potassium phosphate buffer pH 7.0. One unit of NADH oxidase was defined as the amount of enzyme required to oxidise 1 µmol of β -NADH per minute at 30 °C. Assay preparations and dilutions are described in 2.4.1 above. Immobilised NOD preparations were made to volume with 20 mM phosphate buffer (KH $_2$ PO $_4$ -K $_2$ HPO $_4$; pH 7.0). Particle preparations were diluted to within the dynamic range of the kinetic assay (as determined using commercial NOD). No interference from microspheres in the optical pathlength was observed in particle control assays.

2.5. Immobilisation of GDH and NOD

The maximum binding capacity of the microspheres was initially determined using BSA as the model protein (Table 1). These values were used to estimate the approximate loading capacity for GDH and NOD (mg protein to per mg microspheres – dry weight). Aqueous suspensions of GDH (1 mg.ml⁻¹; pH 8.0) and NOD (0.43 mg.ml⁻¹; pH 8.0) were used for immobilisation and were loaded onto the microspheres in excess of the estimated binding capacity (20% and 10% respectively) to ensure particle saturation for determination of binding capacity for these enzymes.

Immobilisation was conducted by incubation of the particle and enzyme suspension for 60 min at 8 °C with end-over-end mixing at 25 rpm (ELMI Intellimixer). The microspheres were recovered by centrifugation at 6000xg for 5 min and the supernatant assayed for unbound protein using the Bio-Rad assay mentioned above. The particles were washed once with 1 ml of MilliQ water. The protein content of the immobilisation and wash supernatants were quantified. To determine the extent of non-specific interaction, the microspheres were further washed with 2 M NaCl to remove non-covalently linked protein, with subsequent protein quantification after microsphere recovery by centrifugation. No loss in enzyme activity was observed for

free NOD or GDH incubated in the immobilisation conditions. Extended incubation times of up to 24 hours did not result in higher immobilisation capacities.

For subsequent immobilisations, the experimentally determined covalent capacity for GDH and NOD was used for microsphere loading. All experimentation was performed in triplicate and data presented as the mean ± standard deviations. The immobilisation capacities of the solid supports are presented as weight of immobilised enzyme per unit solid support (dry weight).

2.6. Enzyme Activity Maintenance

The enzyme activity maintenance was defined as the percentage of the residual activity of the immobilised enzyme, compared to the total enzyme activity loaded (calculated as the difference between the enzyme activity used for immobilisation and the residual activity in solution after removal of the microspheres). These values are expressed as a percentage (Fig. 2). The activity assays for GDH and NOD were performed as described in 2.4.1 and 2.4.2. Specific activities of immobilised enzymes are presented as enzyme activity retained per unit mass of enzyme, as well as per unit mass of enzyme and solid support, including the mass of the non-catalytic solid support to give an indication of the volumetric activity.

2.7. pH Profiling

The pH profile assays were performed using the methods described above in 2.4.1 and 2.4.2. However, the standard assay buffers were replaced with a universal buffer containing 50 mM phosphate (Na₂HPO₄), 50 mM Boric Acid, 33 mM Citric Acid and 50 mM Tris adjusted to pH values between 4-10 using HCl or KOH [28].

2.8. Thermal Stability

Thermal stability was determined by incubation of the free and immobilised GDH preparations at temperatures from 50 to 65 °C with 5 °C increments while NOD was

incubated at 35 and 40 °C. Enzyme activities were assayed initially at 5 min intervals using the methods and conditions described in 2.4.1 and 2.4.2. The time interval was extended for samples displaying high stability. Samples were removed and allowed to cool to room temperature before assaying.

2.9. pH Stability

pH stability was determined by incubation of the free and immobilised GDH and NOD enzyme preparations in universal buffer between pH 2.5 and 10.0. The standard activity assays were performed as described above (2.4.1 and 2.4.2).

3. Results

3.1. Immobilisation of GDH and NOD

The immobilisation of enzymes using aldehyde functional supports occurs through Schiff-base bond formation with the primary amines of the proteins [5]. The immobilisation procedure is performed at pH 8.0 to ensure reactivity of the nucleophilic primary amines with aldehydes available on the microspheres [5]. The microsphere preparations had varying capacities for biomolecule immobilisation, with direct proportionality to the functional group density. This trend was observed when using BSA and NOD for immobilisation (Table 1). However GDH did not appear to follow this trend with ReSyn™ B binding the highest quantity enzyme. This may have been due to a difference in the internal environment and or porosity of this product, which may have induced changes in the nature and/or intensity of interactions between the enzyme and support [29]. Reduction of the resultant Schiff base bonds with cyanoborohydride resulted in a loss of 77% of the enzyme activity (GDH). This may have been owing to the ability of reducing agents to reduce disulphide bridges [30]

3.2. Enzyme Activity Maintenance and Specific Activity of Immobilised GDH and NOD

It is important to determine the maintenance of catalytic activity by the immobilised enzyme when evaluating a technique for immobilisation. The maintenance in activity of GDH and NOD immobilised on the various preparations is shown in figure 2. The enzyme activity maintenance for GDH was generally in the region of 40% with little variance between the microsphere preparations (Fig. 2). Immobilised preparations of NOD displayed activity maintenance with a higher degree of variance ranging from 12.3 to 23.6% (Fig. 2).

The specific activity of an immobilised enzyme is influenced by the nature of the support material, enzyme loading capacity, and maintenance in enzyme activity [13]. The specific activity of immobilised GDH preparations (U.mg⁻¹ enzyme) was lowest on ReSyn[™] C at 42.19 U.mg⁻¹, while ReSyn[™] A and B had higher specific activities of 54.73 and 49.09 U.mg⁻¹ respectively (Table 2). NOD immobilised on ReSyn[™] A had the highest specific activity of 8.09 U.mg⁻¹ followed by ReSyn[™] C with 6.33 U.mg⁻¹ and ReSyn[™] B the lowest at 4.21 U.mg⁻¹ (Table 2). To illustrate the high volumetric activity, the data is further represented as specific activity inclusive of the non-catalytic function (microsphere weight).

3.3. pH Profiling

The pH profile for the free and immobilised GDH and NOD is illustrated in figure 3. All GDH preparations, free and immobilised, showed optimal activity at pH 8.0, while NOD displayed optimal activity at pH 7.0. GDH immobilised on ReSyn™ A exhibited a broader pH profile in the acidic range (pH 5.0-7.0), indicated by a ~2-fold increase in activity in this range compared to that of the native enzyme (Fig. 3). ReSyn™ B and C preparations exhibited the same pH profile as the native enzyme (Fig. 3). There was no shift or notable difference in the pH profile of immobilised NOD preparations when compared to the profile of native NOD (Fig. 3 insert).

3.4. Thermal and pH Stability

GDH immobilised on ReSyn[™] A displayed the highest thermal stability with half-life improvements of 2.89, 6.82, 9.21 and 13.59-fold at 50, 55, 60 and 65 °C respectively over that of the free enzyme (Fig. 4), ReSyn[™] B and C GDH also displayed improved thermal stability, albeit less pronounced (Fig. 4). The enzyme was further stabilised against denaturation under acidic conditions (Table 3a) with ReSyn[™] A again displaying the highest stability with a 2.37 fold improvement in half-life at pH 3.5, and approximately 2 fold at pH 2.5 and 3 (Table 3a). GDH immobilised on ReSyn[™] B and C again displayed reduced pH stability compared to ReSyn[™] A. These findings appear to confirm that functional stability is proportional to the reactive group density used for immobilisation [5, 31].

Incubation of immobilised NOD at elevated temperatures indicated that only ReSyn™ A and B provided improved thermal and pH stability. ReSyn™ C did not display any improved thermal stability over the free enzyme during incubation at 40 and 45 °C (Table 3b). ReSyn™ A provided improved protection against thermal denaturation of NOD with improvements at 40 °C (1.87 fold) and 45 °C (2.25 fold) respectively. However, this did not coincide with pH stability. ReSyn™ B provided the best enhancement in pH stability, increasing the half-life of NOD activity from 14 to 141 minutes at pH 3.0. ReSyn™ A and C appeared to marginally enhance the stability at pH 3.0 (Table 3b).

4. Discussion

The claimed functional group densities ranging from 300 to 3500 µmoles.g⁻¹ (ReSyn™ Biosciences) and the fibrous interpenetrating network of ReSyn™ microspheres [14] are proposed to provide an increased surface area resulting in a high binding capacity (www.resynbio.com). This feature was witnessed for the immobilisation of the standard control protein (BSA), GDH and NOD. The results presented here for ReSyn™ microspheres far exceed previously reported literature. Covalent immobilisation of GDH on controlled pore silica by Baron *et al.* [20] demonstrated a maximum binding capacity of 0.4 mg.g⁻¹ (Table 4) while here we report a loading capacity of 820 mg.g⁻¹ by ReSyn™ B. The lack of literature on GDH immobilisation is surprising considering its value as a co-factor recycling agent in

biocatalysis, but may be associated with the challenges of immobilising multimeric enzymes [25, 26]. The ReSyn™ microspheres further resulted in improved enzyme activity maintenance of 41.5% for ReSyn™ A, as compared to the values reported by Baron *et al.* [20] of 1.15% (Table 4). The resultant specific activities of ReSyn™ A immobilised GDH was calculated as 54.73 U.mg⁻¹ (or 33.22 U.mg⁻¹ inclusive of microsphere support) through a combination of relatively high enzyme activity maintenance and the high capacity of the microspheres. In comparison, the specific activity for the immobilisation on silica support was previously reported as 2.74 U.mg⁻¹ (enzyme only) (Table 4).

The binding capacity achieved for the immobilisation of NOD on aldehyde functional ReSyn™ also exceeded previously reported results. Sanjust *et al.* [32] reported a maximum capacity of 17 mg.g⁻¹ for the covalent immobilisation of NOD on both cyanogen bromide activated PVA beads and Sephacryl S-200 HR treated with 2,4,6-Trichloro-1,3,5-triazine (TCT) (Table 4) while in this study ReSyn™ A achieved 1 288 mg.g⁻¹. More recently, NOD was immobilised on single-walled carbon nanotubes (SWCNTs) with a resultant capacity of 470 mg.g⁻¹ [33]. Although this is lower than what was achieved on ReSyn™, the SWCNT achieved an enzyme activity maintenance of 92% ([33]; Table 4) while the highest achieved for ReSyn™ was 23.6 %. This comparatively low maintenance in enzyme activity maintenance resulted in a specific activity of 8.09 as compared to 52.4 U.mg⁻¹ obtained with SWCNT's.

Apart from the well understood factors that may have affected enzyme activity maintenance of NOD and GDH such as steric hindrance [31, 34, 35]; interaction and orientation on the support and change in the micro-environment upon immobilisation [8, 13]; the high capacity of ReSyn™ microspheres has a high potential for mass transfer limitations of substrate and/or products [13, 29, 34, 36, 37].

More specifically, for GDH the loss in activity may be as a direct result of the active site lysine residue (Lys-201) taking part in the immobilisation process, an integral amino acid required for activity and/or ligand binding in the enzyme [38]. ReSyn™ microspheres are loosely-linked polyethyleneimine matrices (www.resynbio.com). Interestingly, PEI has been shown to play a role in the stabilisation of oxygen-

sensitive nitrilases, which was attributed to the low permeability and reduced solubility of oxygen in this polymer [39]. Since NOD utilises oxygen as a substrate, this property of polyethyleneimine may have resulted in the relatively low enzyme activity maintenance of 23.6%.

It is well known that the microenvironment of the immobilisation matrix can result in a shift in the optimal pH of an enzyme [40, 41]. In this experimentation there appeared to be no change in the optimum pH of either of the immobilised enzymes (Fig. 3). However, ReSyn™ A immobilised GDH displayed enhanced activity in the acidic pH range, potentially enabling applications in these conditions where this enzyme is currently not suitable. An example of where this feature may enhance applicability of this enzyme is the synthesis of gluconic acid in the food industry [42].

5. Conclusions

In general, ReSyn™ immobilised enzymes displayed improved stability at elevated temperature and acidic pH conditions. For immobilised enzyme preparations the highest thermal stability was displayed on ReSyn™ A microspheres having the highest functional group density of 3 500 µmol.g⁻¹. This improved stability of immobilised enzymes is proposed to result from "rigidification" of the enzyme's threedimensional structure which provides resistance to conformational change induced by environmental conditions [5, 31, 43, 44], while the degree of multipoint attachment (a feature of the available functional group density) has been proposed to be directly proportional to improvement in stability [5, 31]. The results presented here are in good agreement with this hypothesis. Stability in immobilised enzymes may also be enhanced by the micro-environment provided by the support [5, 12, 31]. The proposed structure of the microspheres, i.e. a porous loosely-linked polymer matrix, appears to allow shielding from the harsh external environment. The results demonstrate a microsphere with an exceptional high protein binding capacity with maximum loading of 50% achieved for GDH and 129% for NOD. Of particular interest is the immobilisation of the multimeric enzyme GDH since multi-subunit enzymes are notoriously difficult to immobilise [25, 26]. The successful immobilisation of these enzymes may be used to enhance their suitability as cofactor recycling agents in biocatalytic applications.

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List of Tables

Table 1: Binding capacity of GDH and NOD immobilised on various ReSyn™
 preparations

ReSyn™	Microsphere dry	Binding Capacity (mg.g ⁻¹ dry support)			
Preparation	weight (mg.ml ⁻¹)	BSA	GDH	NOD	
Α	8.56 ± 0.61	1 711 ± 20	508 ± 2	1 288 ± 1	
В	6.35 ± 0.35	950 ± 40	820 ± 2	837 ± 3	
С	7.41 ± 0.40	292 ± 30	163 ± 30	220 ± 12	

Table 2: Specific activities of GDH and NOD calculated including and excluding the non-catalytic load (support)

ReSyn™	Specific activity (U.mg ⁻¹ enzyme)		Specific activity (U.mg ⁻¹ support and enzy	
Preparation	GDH	NOD	GDH	NOD
Α	54.73 ± 0.06	8.09 ± 0.73	33.22 ± 0.01	6.75 ± 0.03
В	42.19 ± 0.23	4.21 ± 0.08	33.01 ± 0.07	1.80 ± 0.05
C	49.09 ± 0.22	6.33 ± 0.06	12.76 ± 0.03	4.95 ± 0.03

The specific activities of commercial enzymes were determined as 116 U.mg⁻¹ for GDH and 34.2 U.mg⁻¹ for NOD.

Table 3a: Half-life values indicating pH stability of free and immobilised GDH

	pH stability (t _{50%} -min)		
Preparation	2.5	3	3.5
Native GDH	8 ± 1.61	27 ± 1.02	140 ± 2.10
ReSyn™ A-GDH	18 ± 2.10	50 ± 2.32	332 ± 2.87
ReSyn™ B-GDH	7 ± 1.59	34 ± 1.70	274 ± 1.01
ReSyn™ C-GDH	7 ± 1.81	28 ± 0.80	239 ± 1.21

Table 3b: Half-life values for free and immobilised NOD obtained from thermal and pH stability assays

	Thermal stability (t _{50%} - min)		pH stability (t _{50%} - min)		
Preparation	40 °C	45 °C	3.0	4.0	10.0
Native NOD	47 ± 1.77	40 ± 1.84	14 ± 1.74	114 ± 0.61	116 ± 0.47
ReSyn™ A-NOD	88 ± 1.02	90 ± 2.38	48 ± 0.32	173 ± 0.45	164 ± 0.406
ReSyn™ B-NOD	44 ± 1.70	31 ± 1.81	141 ± 3.53	265 ± 1.16	233 ± 0.79
ReSyn™ C-NOD	43 ± 0.34	35 ± 0.93	43 ± 2.47	113 ± 1.94	147 ± 1.52

Table 4: Covalent immobilisation of GDH and NOD on various supports

Enzyme	Immobilisation Support	Binding Capacity (mg.g ⁻¹ support)	Maintenance in Activity (%)	Specific Activity (U.mg ⁻¹ _{enzyme})	Reference
GDH	CPS ^a -500	0.4	1.15	2.74	[20]
	ReSyn™ A	508	41.5	54.73	Current Study
NOD	Cross-linked PVA (CNBrb)	16.5	21	4.3	[32]
	Sephacryl S-200 HR (TCT ^c)	16.5	23	4.9	[32]
	Glutarylaminopropyl-PVAd (NHSe)	16	38	8.3	[32]
	SWCNTs ^f	470	92	54.2	[33]
	ReSyn™ A	1 288	23.6	8.09	Current Study

a - controlled pore silica, b - cyanogen bromide, c - 2,4,6-Trichloro-1,3,5-triazine, d - polyvinyl alcohol, e - N-hydroxy-succinimide, f - single walled carbon nanotubes

482 **List of Figures** 483 484 Fig. 1: Enzymatic reactions for (a) glucose dehydrogenase and (b) NADH oxidase 485 indicating the utilisation and regeneration of the nicotinamide co-factors. 486 487 Fig. 2. Average maintenance in activity (as defined in the text) displayed by GDH (■) 488 and NOD (■) immobilised on various ReSyn™ preparations. The data are presented 489 as the mean of triplicate data \pm the standard deviation. 490 Fig. 3. pH profile of native GDH (→) and GDH immobilised on ReSyn[™] A (- → -), 491 492 B (--▲--) and C (----♦---). Optimal activity was displayed at pH 8.0. ReSyn™ B-GDH 493 displayed a broader pH profile with increased activity over the acidic range. Insert: 494 Free (→) and immobilised NOD on ReSyn™ A (- → -), B (- · • -) and C (· · · • · ·), 495 optimal activity was displayed at pH 7.0 for all preparations. The data are presented 496 as the mean of triplicate data \pm the standard deviation. 497 498 Fig. 4. Temperature stability of non-immobilised (•) and GDH immobilised on ReSyn™ A (•), B (▲) and C (♦) during incubations at 50, 55, 60 and 65 °C. Insert: 499 500 Half-life plots of native GDH (→) and GDH immobilised on various ReSyn™

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