

# **SOLAR SELECTIVE ABSORBER FUNCTIONALITY OF CARBON NANOPARTICLES EMBEDDED IN $\text{SiO}_2$ , $\text{ZnO}$ and $\text{NiO}$ MATRICES**

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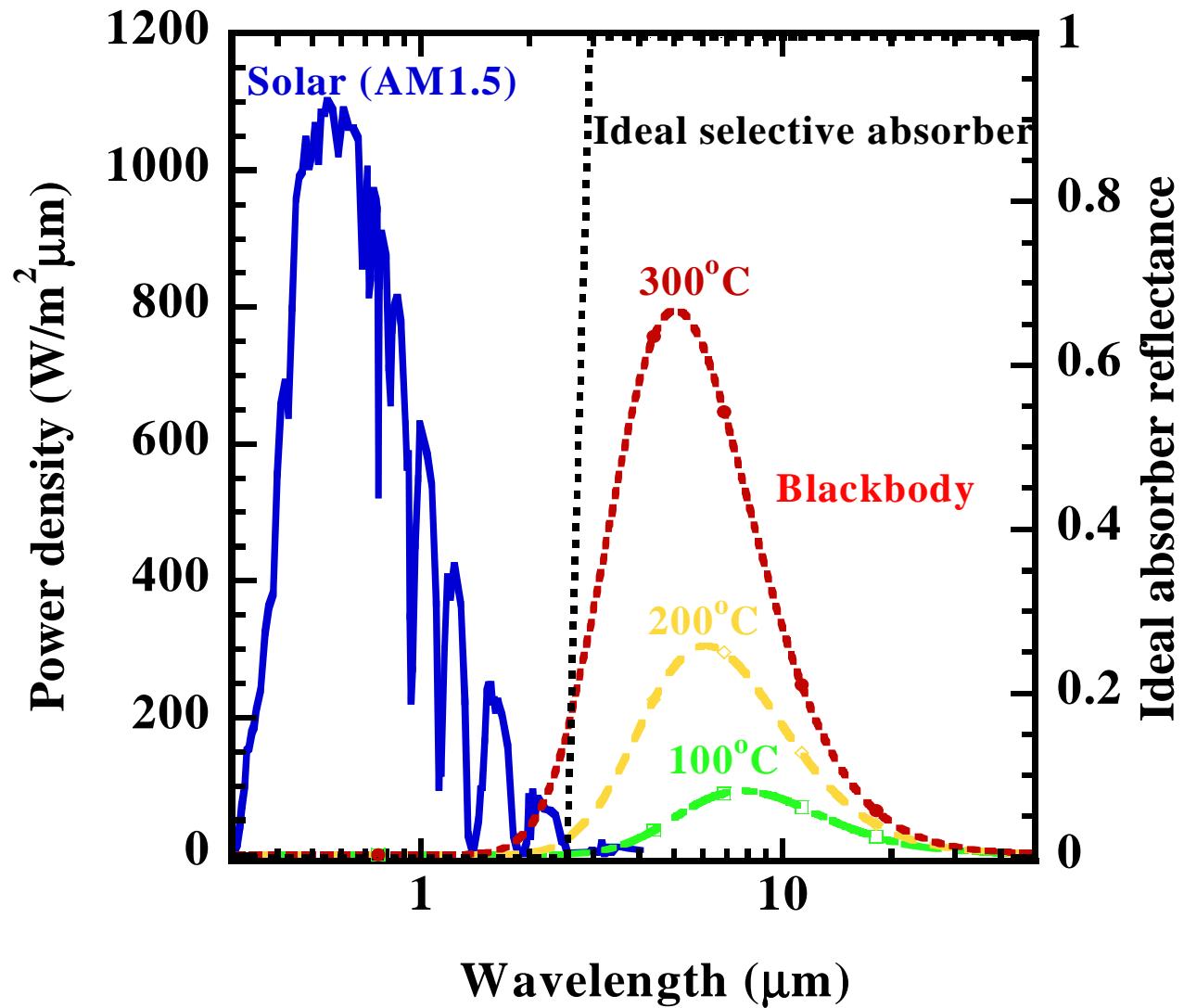
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# Concept and motivation





# Electromagnetic wave propagation

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- Interaction of light with matter.
- Complex refractive index.

$$E_m = E_0 \exp(-k\alpha x / c) \exp[i(\omega t - n\alpha x / c)]$$

- Magnetic field is not altered in optical and IR.
- Logarithmic reduction in intensity for a film thickness  $d$  is  $\ln(I/I_0) = -\alpha d = -4\pi k d/\lambda$ .



## Tailoring materials

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- Factor  $kd/\lambda$  determines the efficiency of a solar absorber surface.
- Proper combination of  $k$  and  $d$ .
- Homogeneous films – can only vary  $d$ .
- Composite films – change  $k$ ! Practically constant for most materials used.
- Can also tune  $n$  by changing porosity.



## Tailoring materials

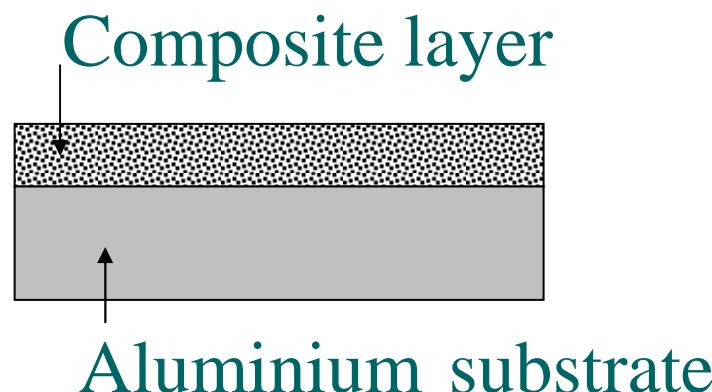
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- Therefore can only tune  $d$  to place the crossover at an appropriate wavelength.
- Reflectance and efficiency of absorber surface depends on  $k$  and  $d$ .
- Near-normal reflectance of bulk material is:  
$$R = \left| \frac{1 - N}{1 + N} \right|^2$$

# Device structure

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- Many designs are possible.
- A tandem device: a composite layer deposited on a metallic substrate.





# Effective medium approximations

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$$\epsilon = N^2$$

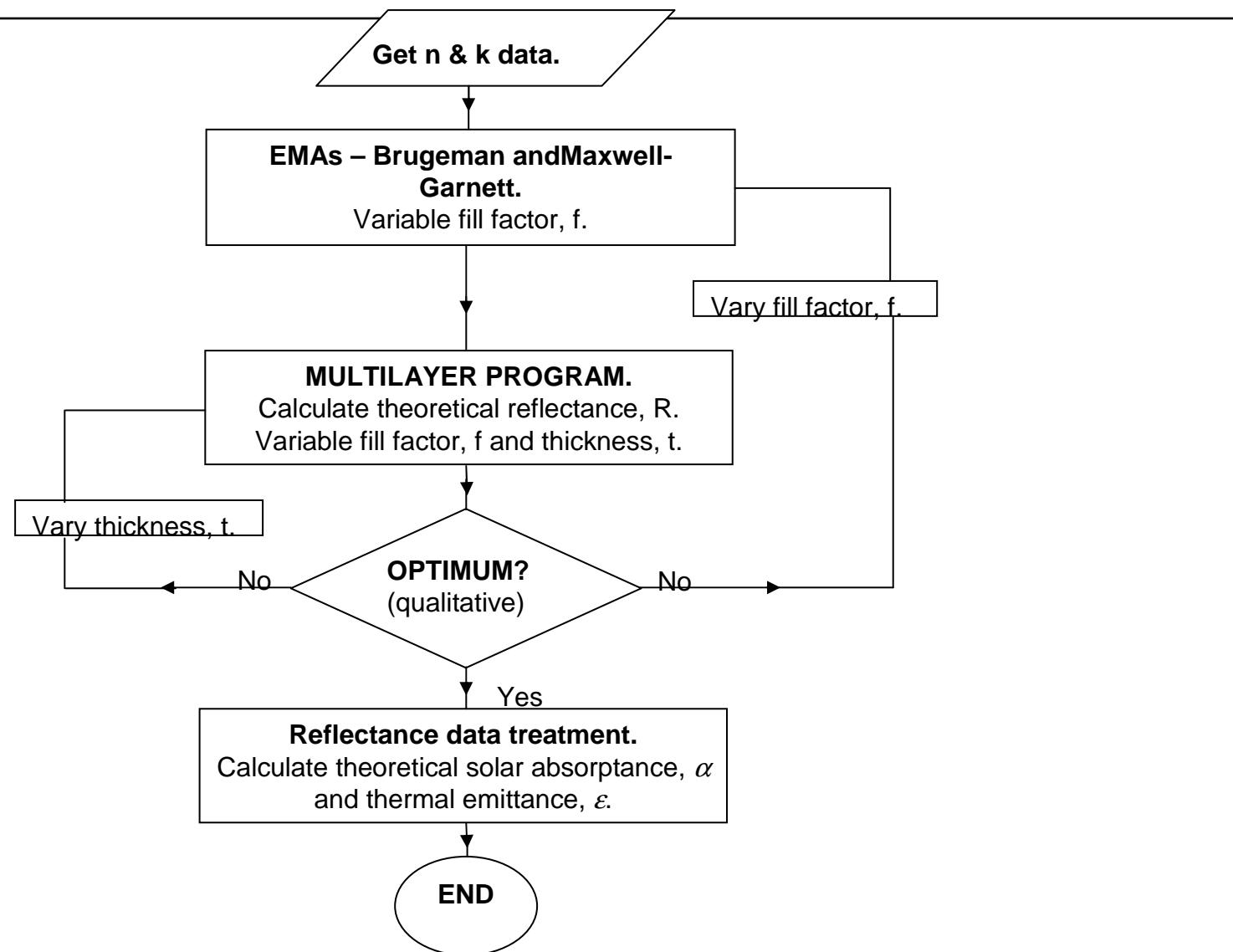
- Bruggeman:

$$f_a \frac{\epsilon_a - \epsilon_{Br}}{\epsilon_a + 2\epsilon_{Br}} + (1-f_a) \frac{\epsilon_b - \epsilon_{Br}}{\epsilon_b + 2\epsilon_{Br}} = 0$$

- Maxwell-Garnett:

$$\epsilon_{MG} = \epsilon_b \frac{\epsilon_a + 2\epsilon_b + 2f_a(\epsilon_a - \epsilon_b)}{\epsilon_a + 2\epsilon_b - 2f_a(\epsilon_a - \epsilon_b)}$$

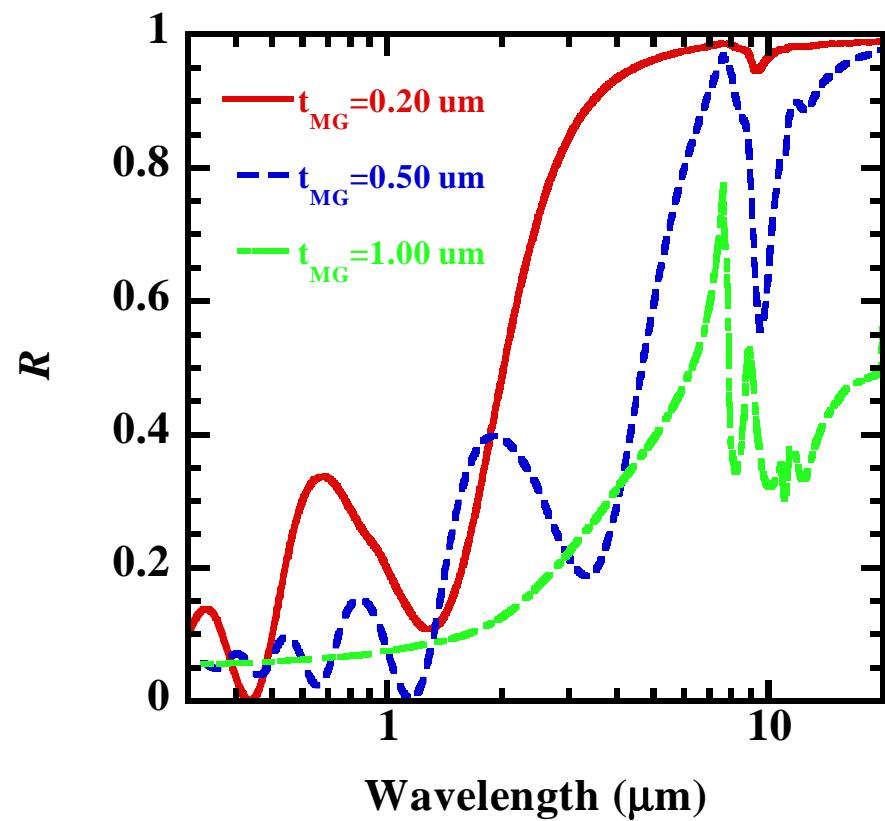
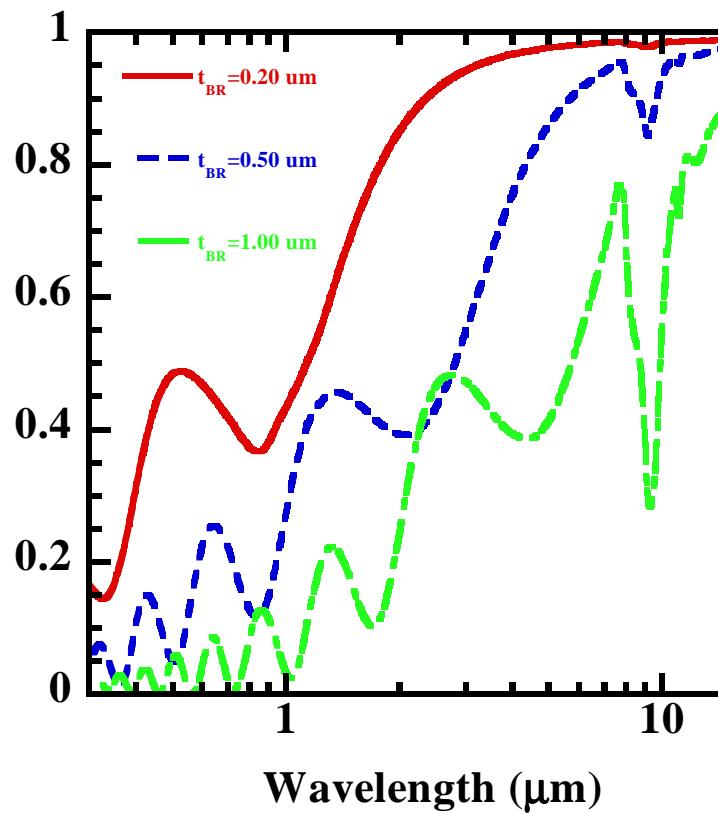
# Theoretical optimisation



# Theoretical reflectance

○ Bruggeman

○ Maxwell-Garnett



# Experimental procedure

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- Sol production



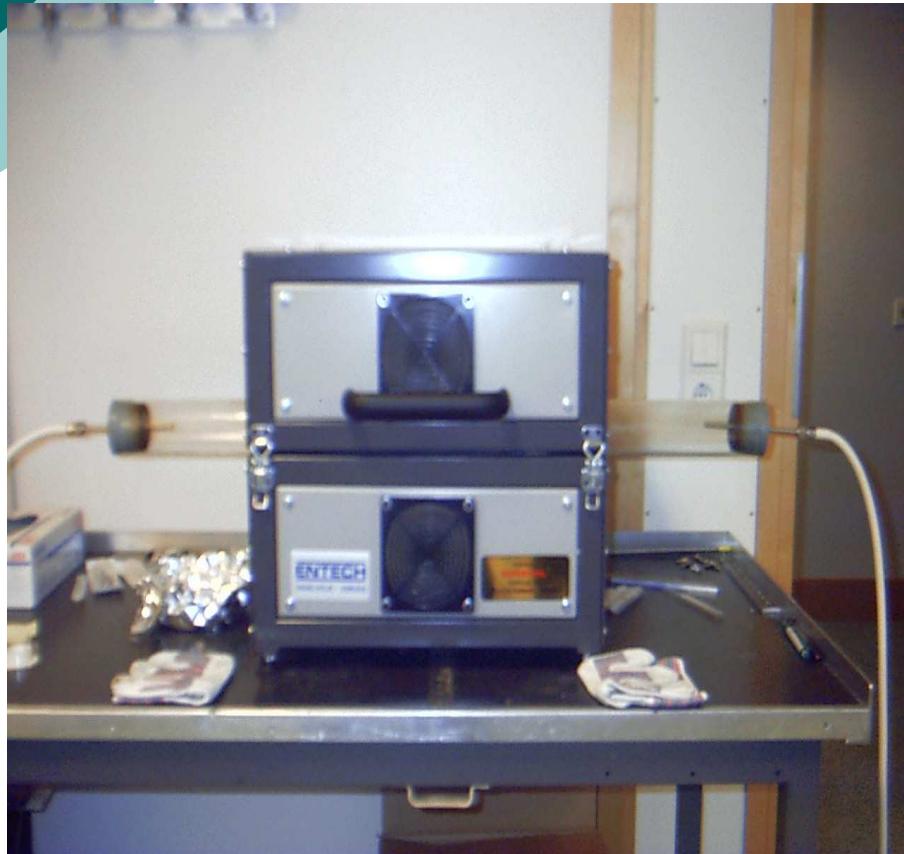
- Tube furnace



# Experimental procedure

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- Carbonization



- Filtration of nitrogen gas



# Optical characterisation

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- Varian Cary 500
- UV-VIS-NIR
- Bomem DA8
- NIR-IR





## Non-optical characterisation

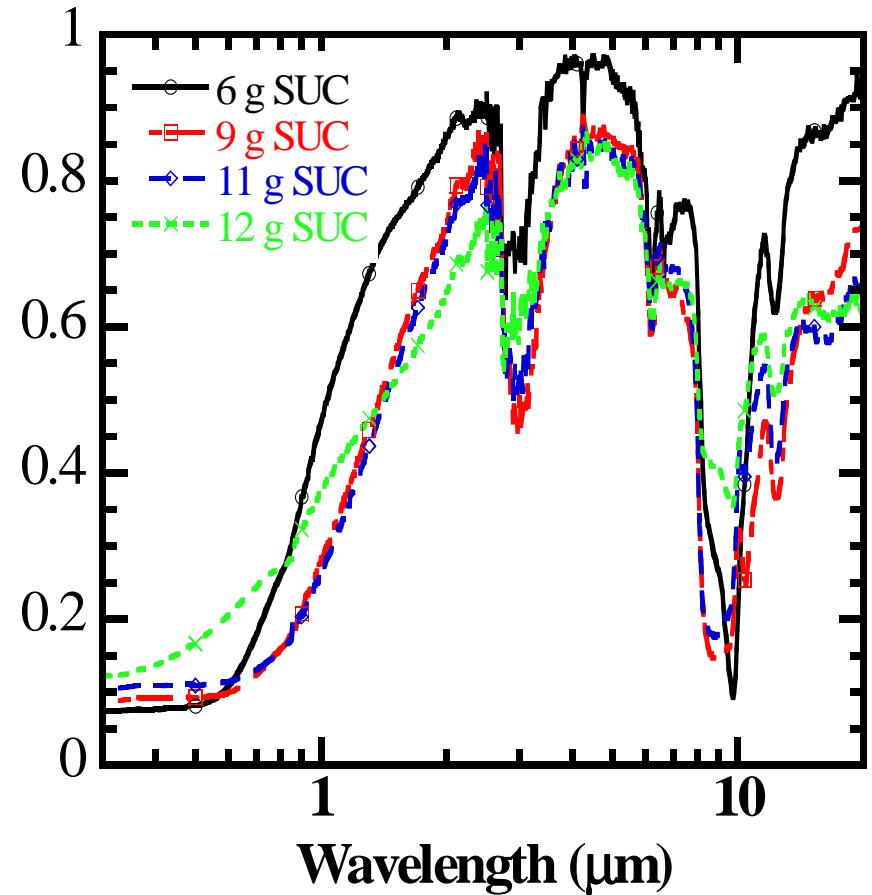
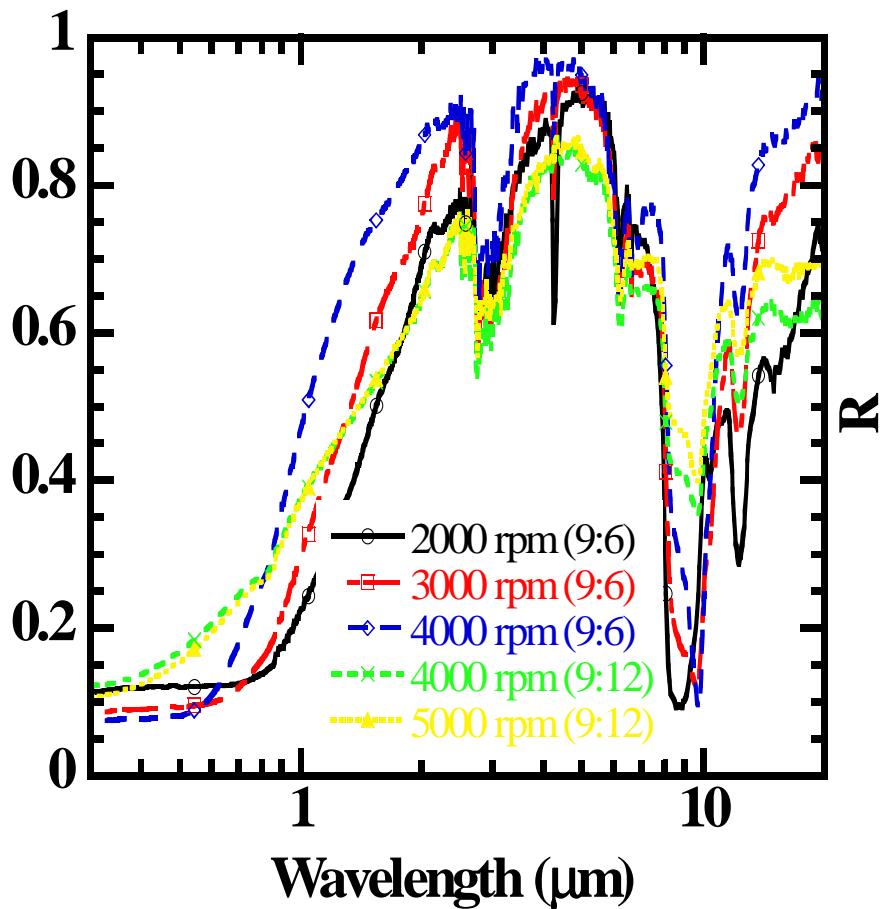
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- SEM: Philips XL30  
Surface morphology
- XRD: Philips PW1840 Diffractometer  
 $\text{Cu K}_\alpha$ , 35 keV, 30 mA  
Crystal structure

# Experimental results: $\text{SiO}_2$ samples

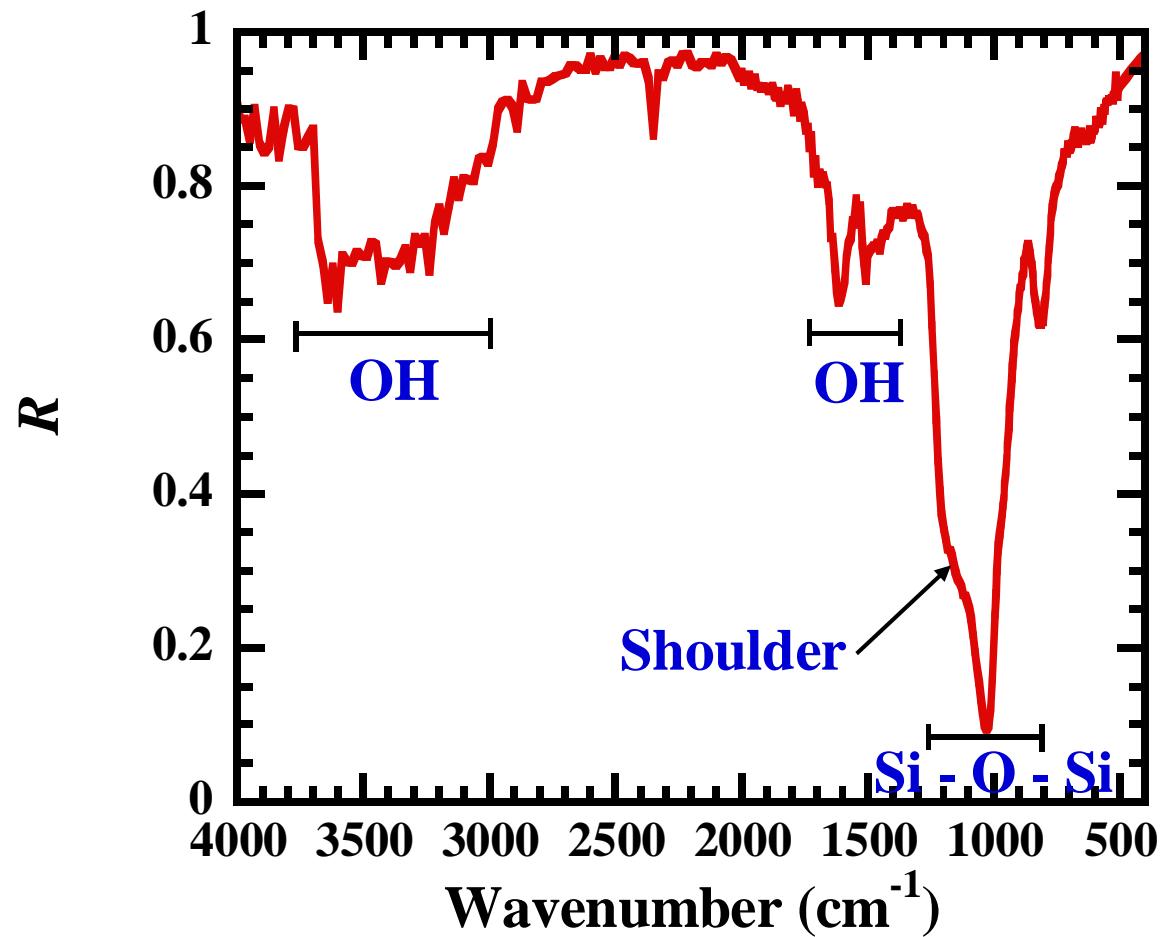
○ Spin-coating speed

○ Carbon precursor



# Experimental results: IR spectrum

- FTIR reflectance spectrum



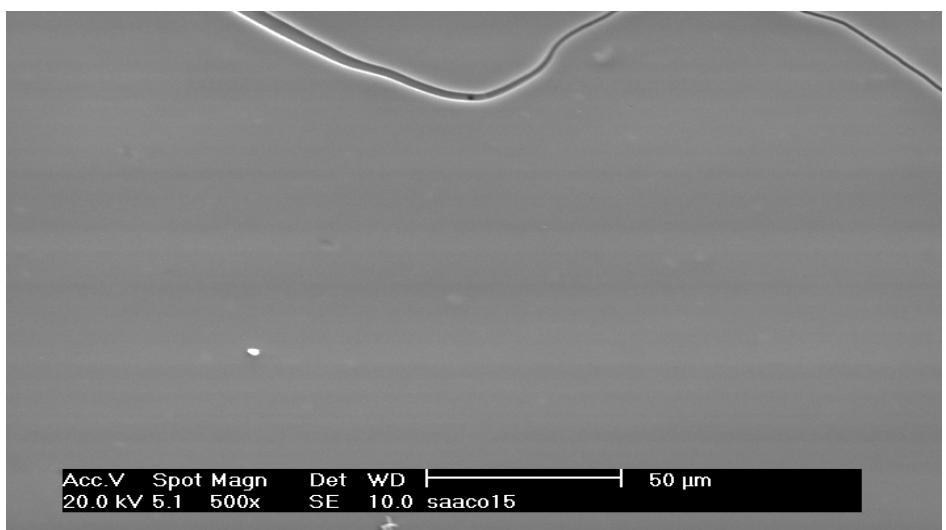
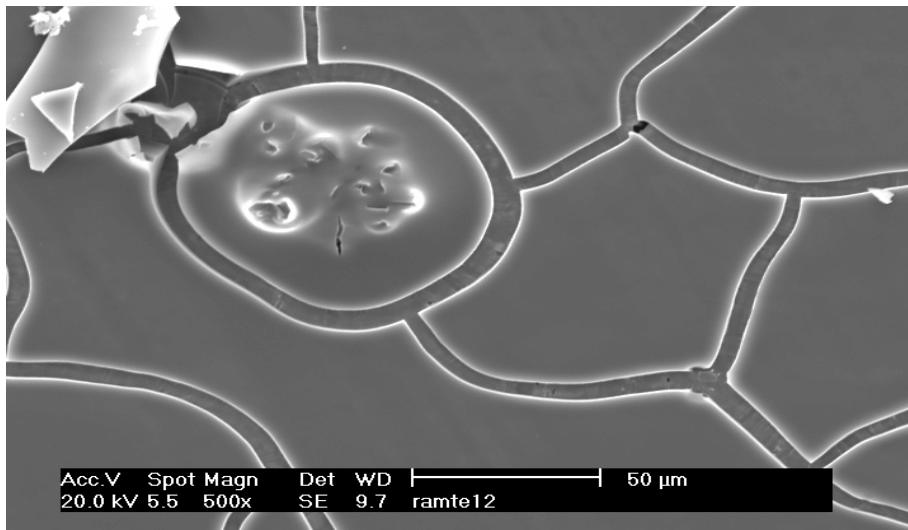
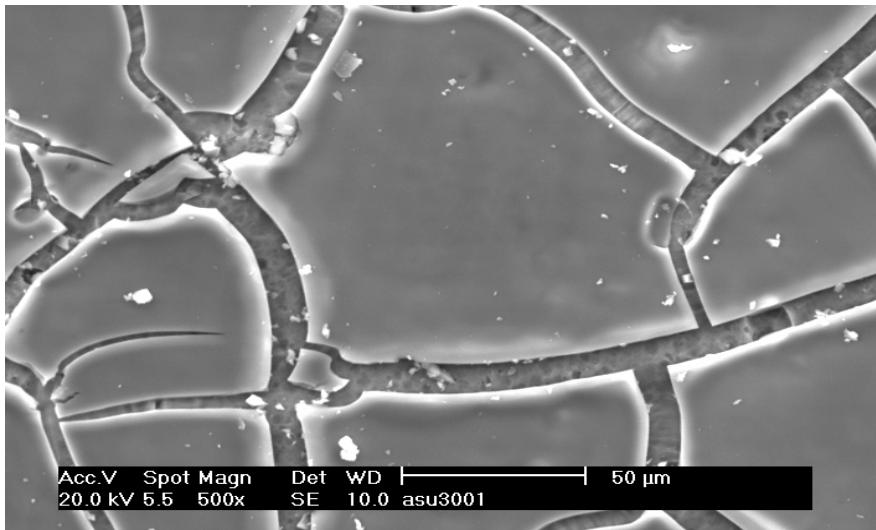


## Experimental results: defects

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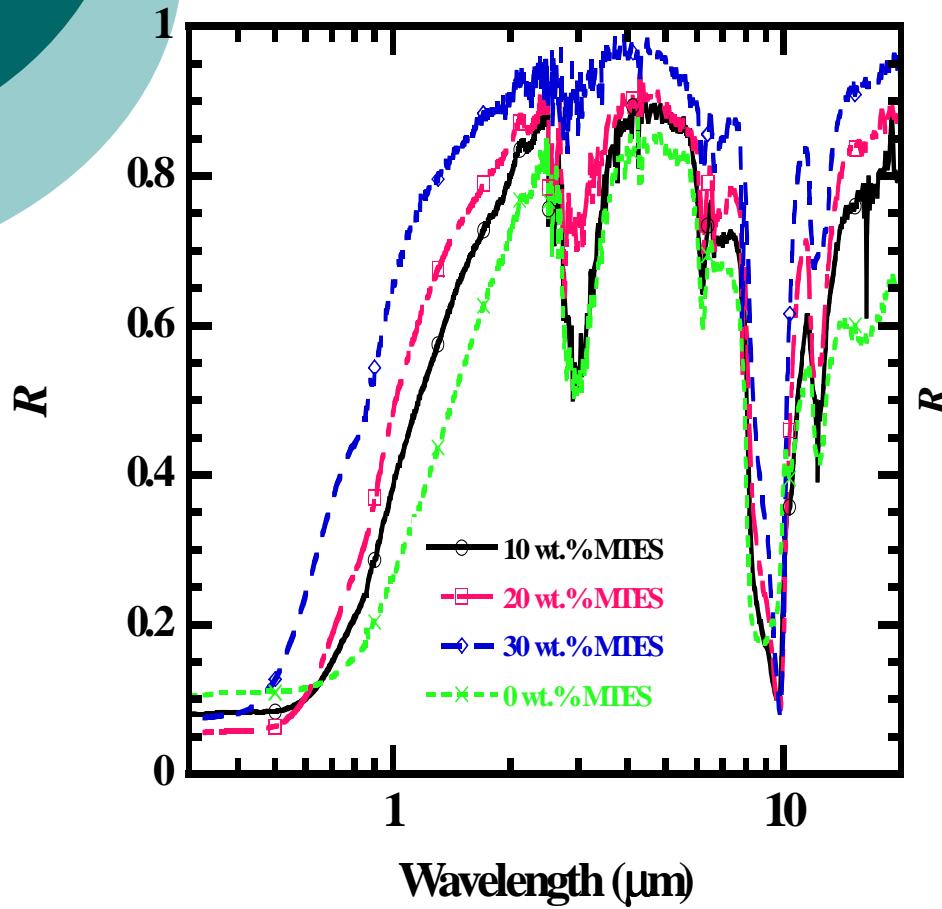
- Problem- cracked films
- Solutions to cracking
- MTES &  $\text{Ac}_2\text{O}$

# Experimental results: defects

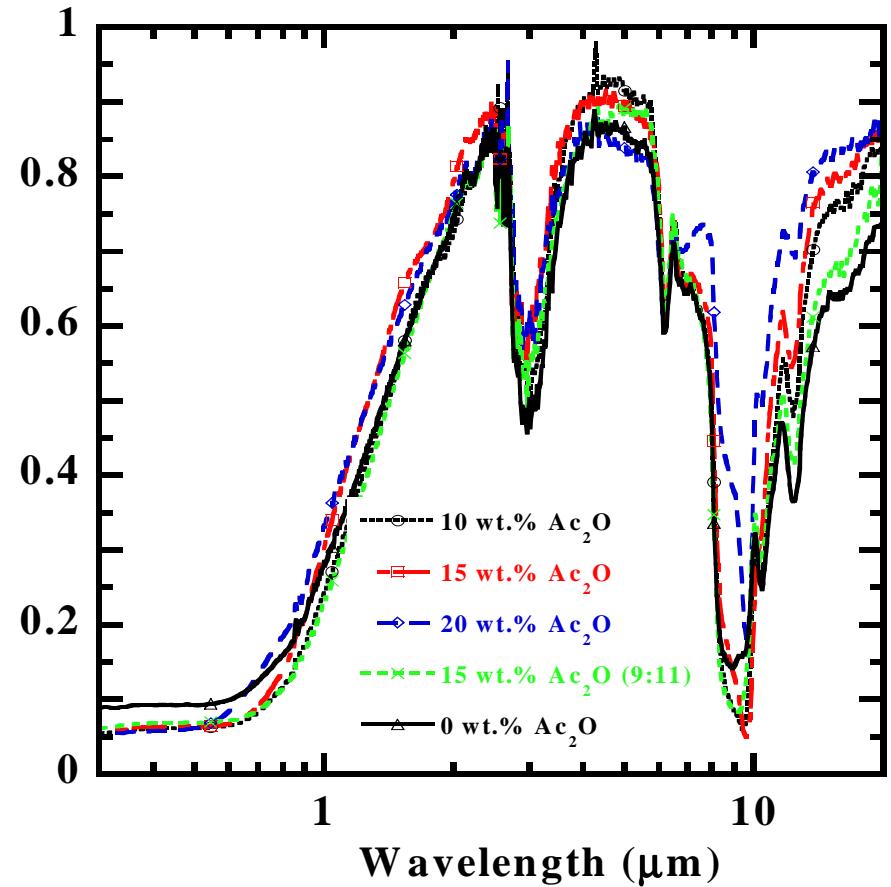


# Experimental results: spectra

○ Addition of MTES

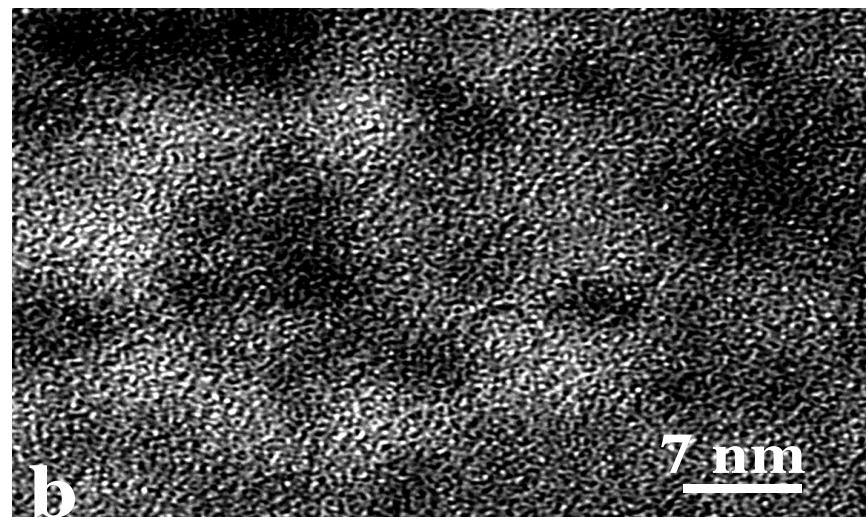
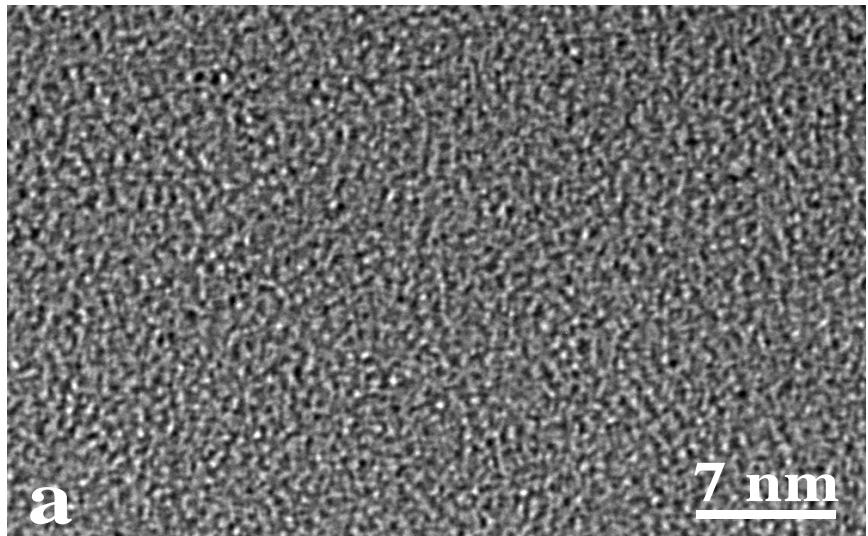
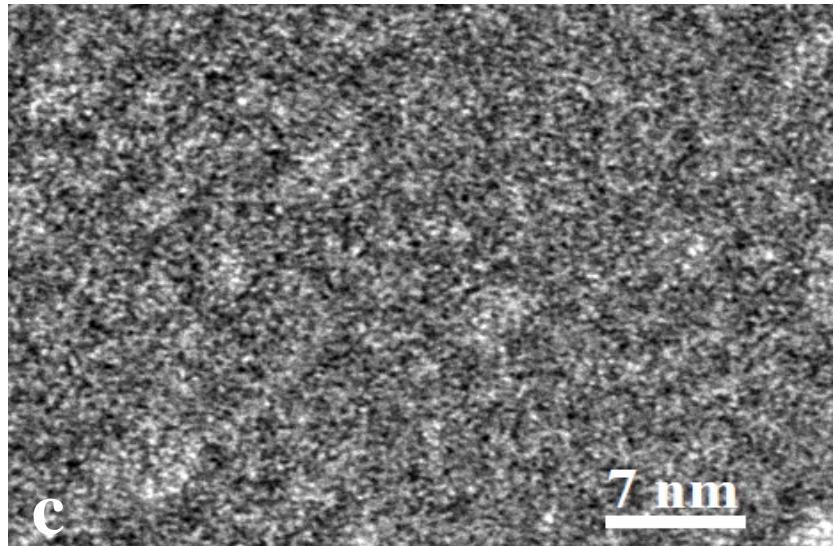


○ Addition of  $\text{Ac}_2\text{O}$



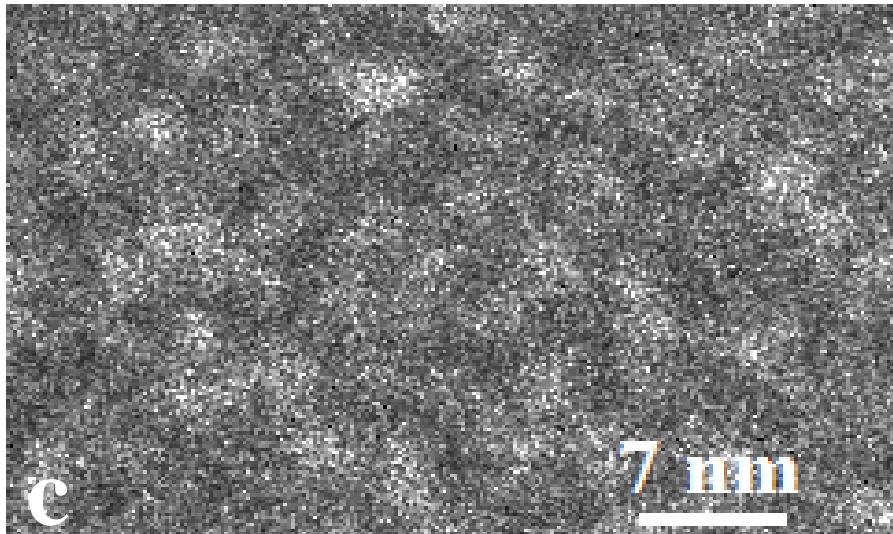
# Experimental results: X-HRTEM

- (a) TEOS only
- (b) TEOS + Ac<sub>2</sub>O
- (c) TEOS + MTES
- Uniform distribution
- Segregation



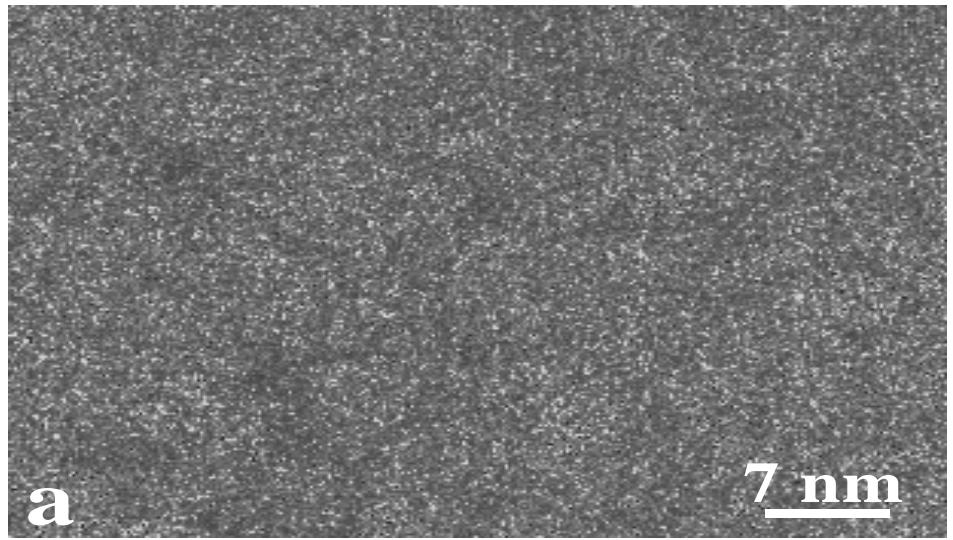
# Experimental results: EELS

- EELS mapping
- Carbon K peak
- (a) TEOS only
- (b) TEOS + Ac<sub>2</sub>O
- (c) TEOS + MTES



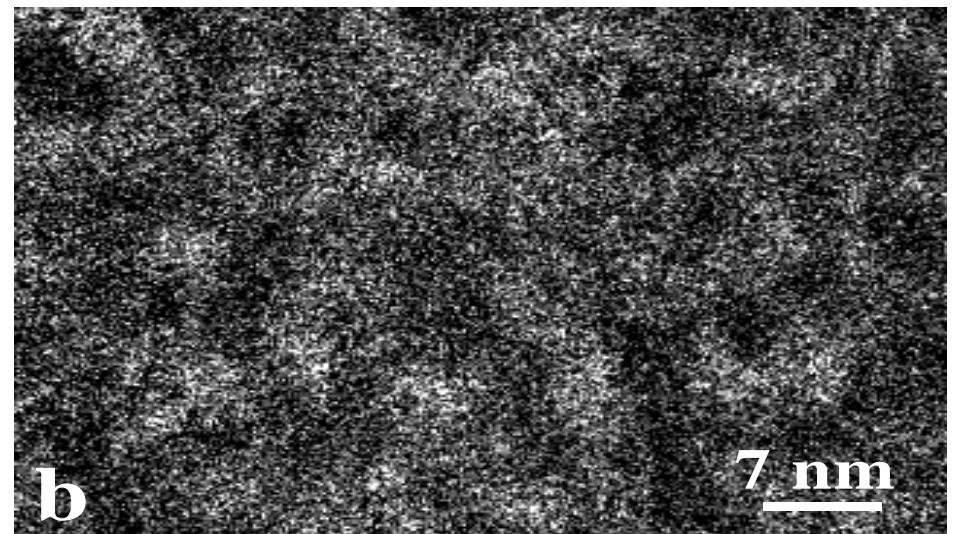
c

7 nm



a

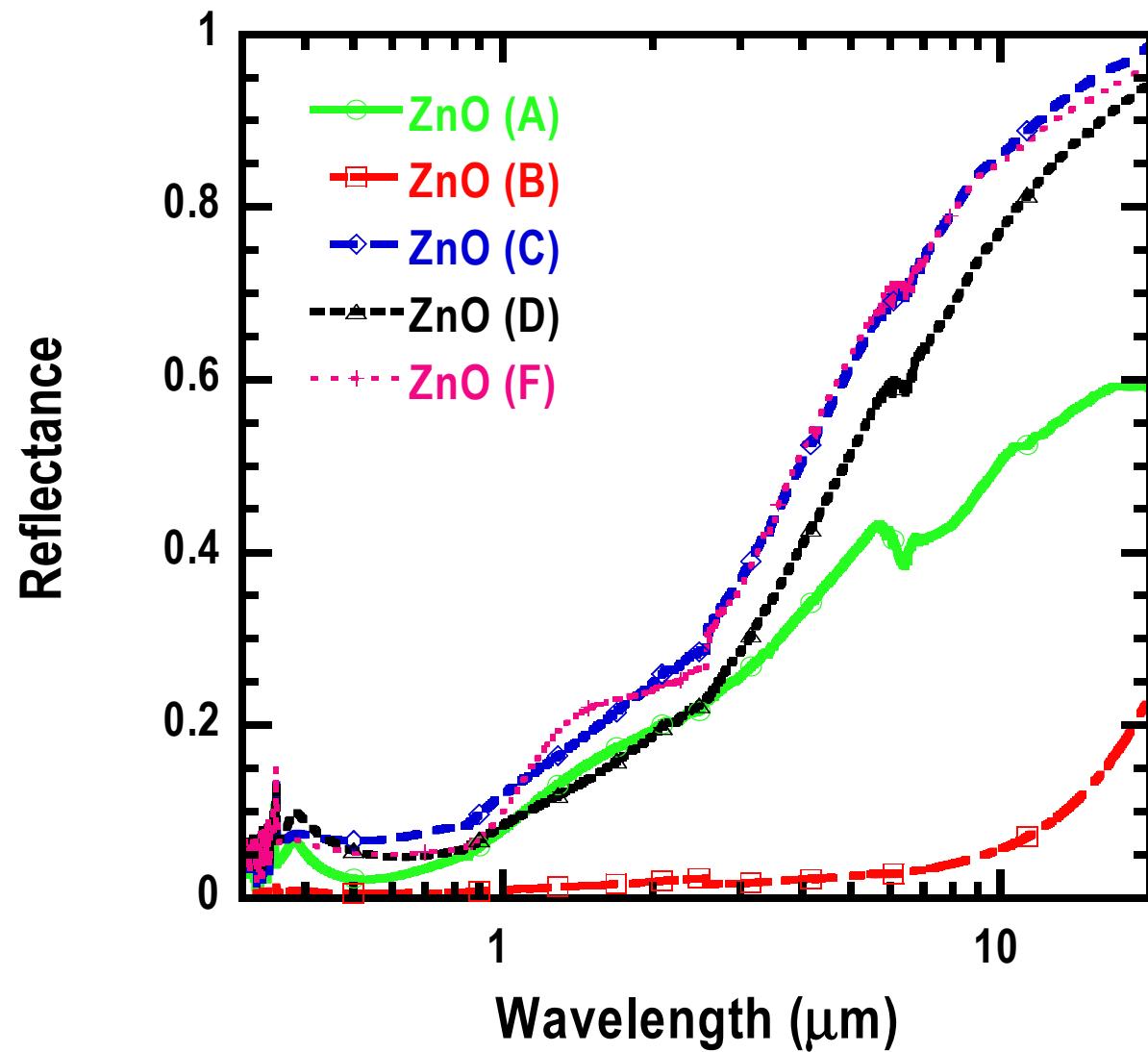
7 nm



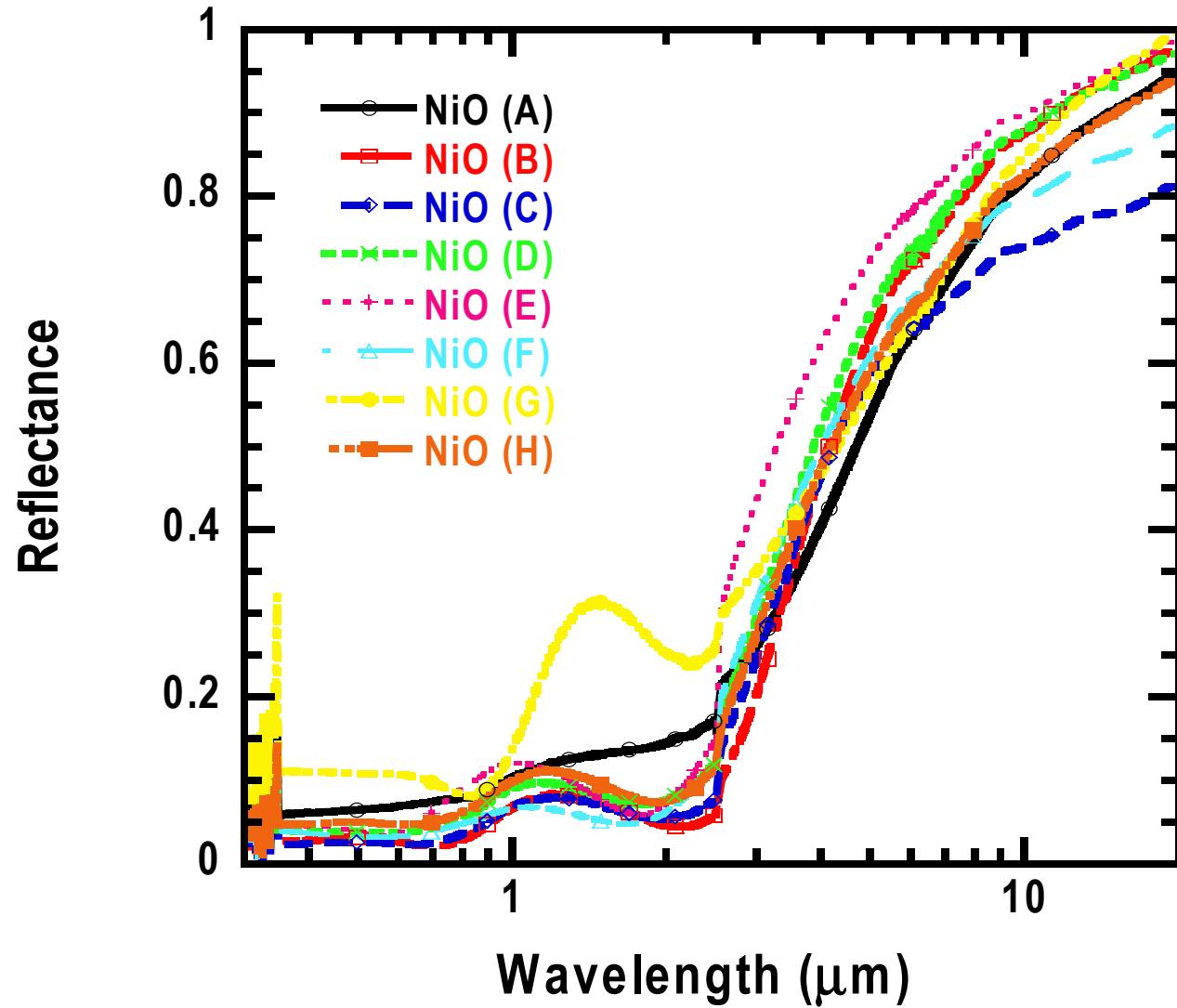
b

7 nm

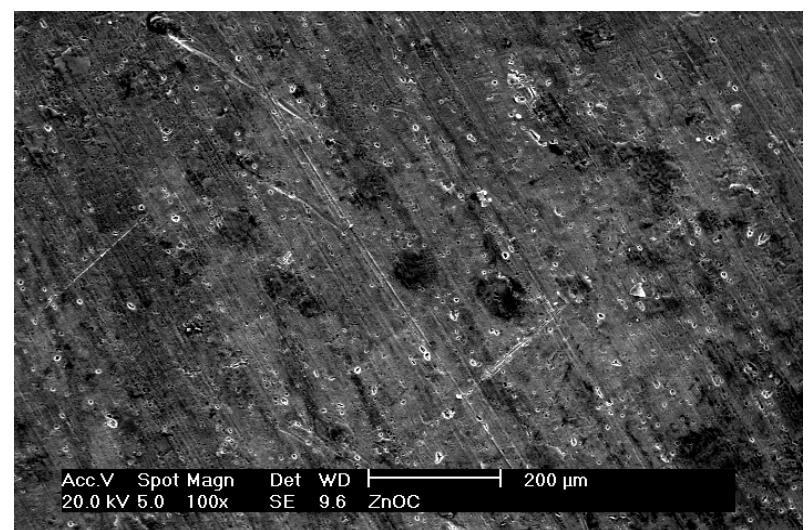
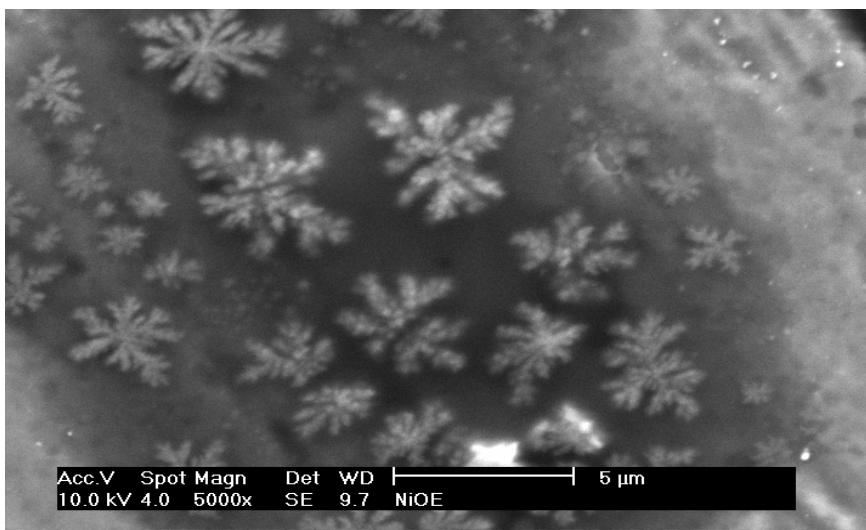
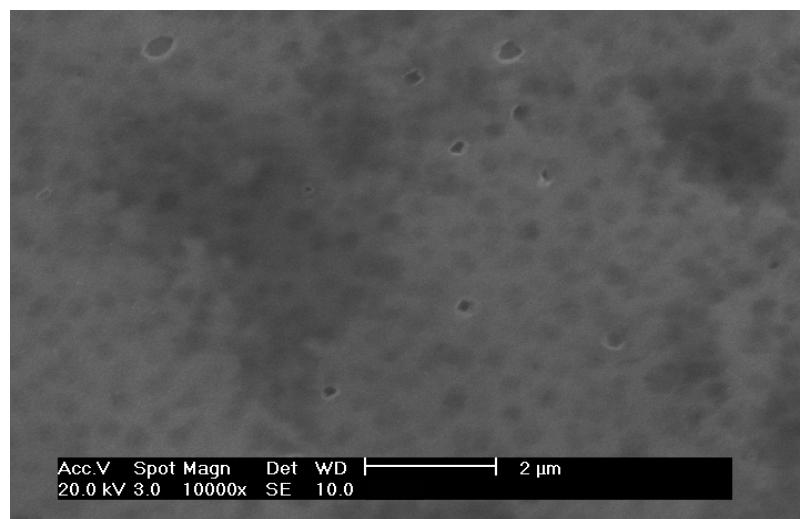
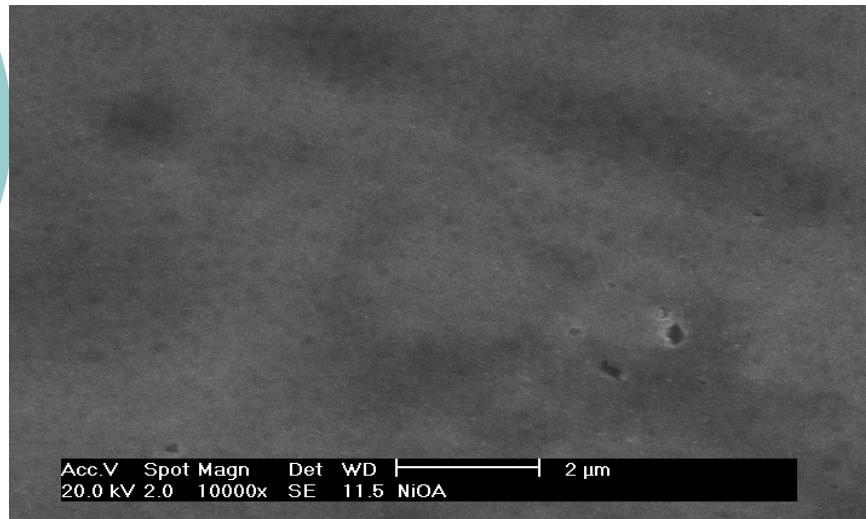
# Experimental results: ZnO samples



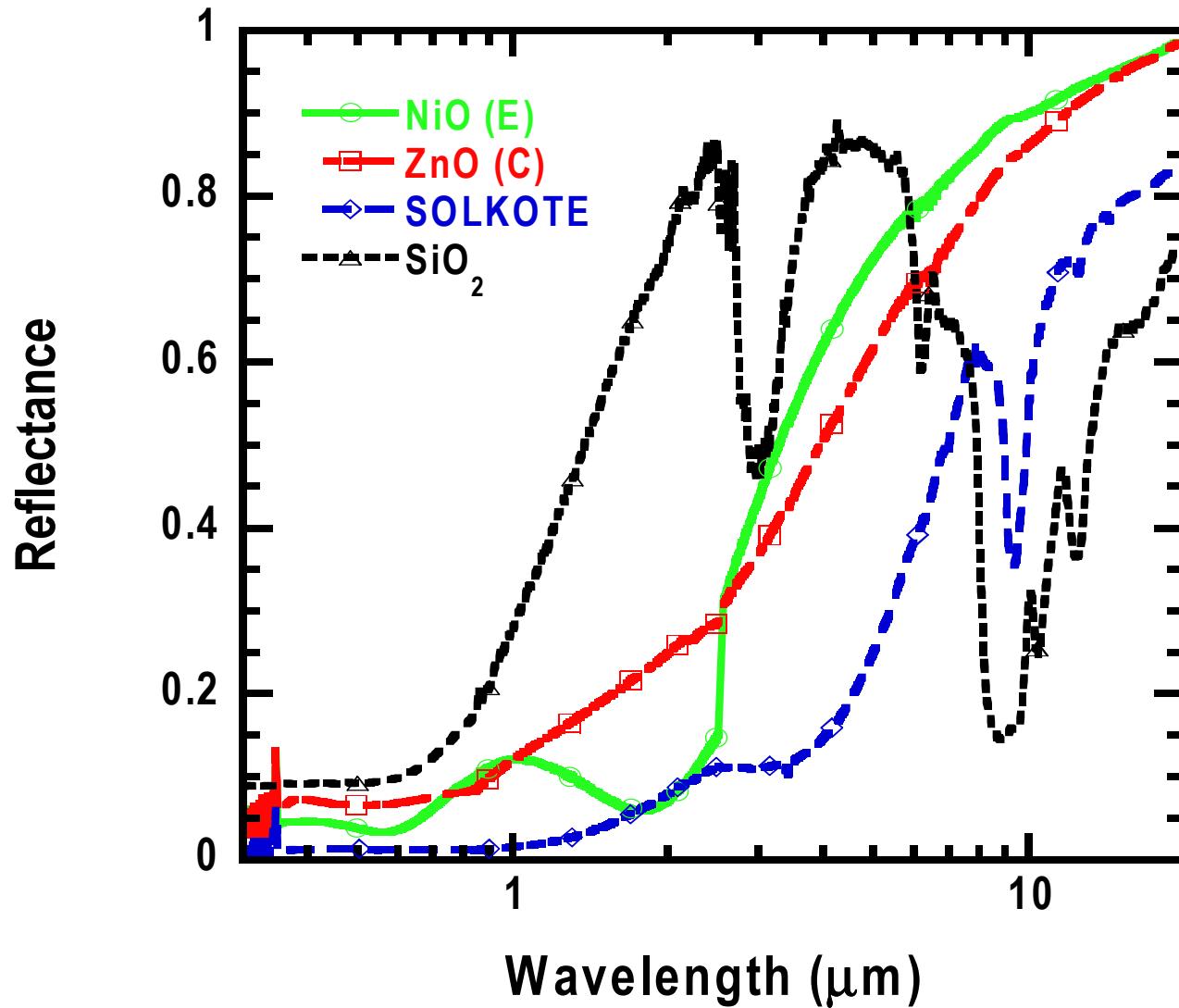
# Experimental results: NiO samples



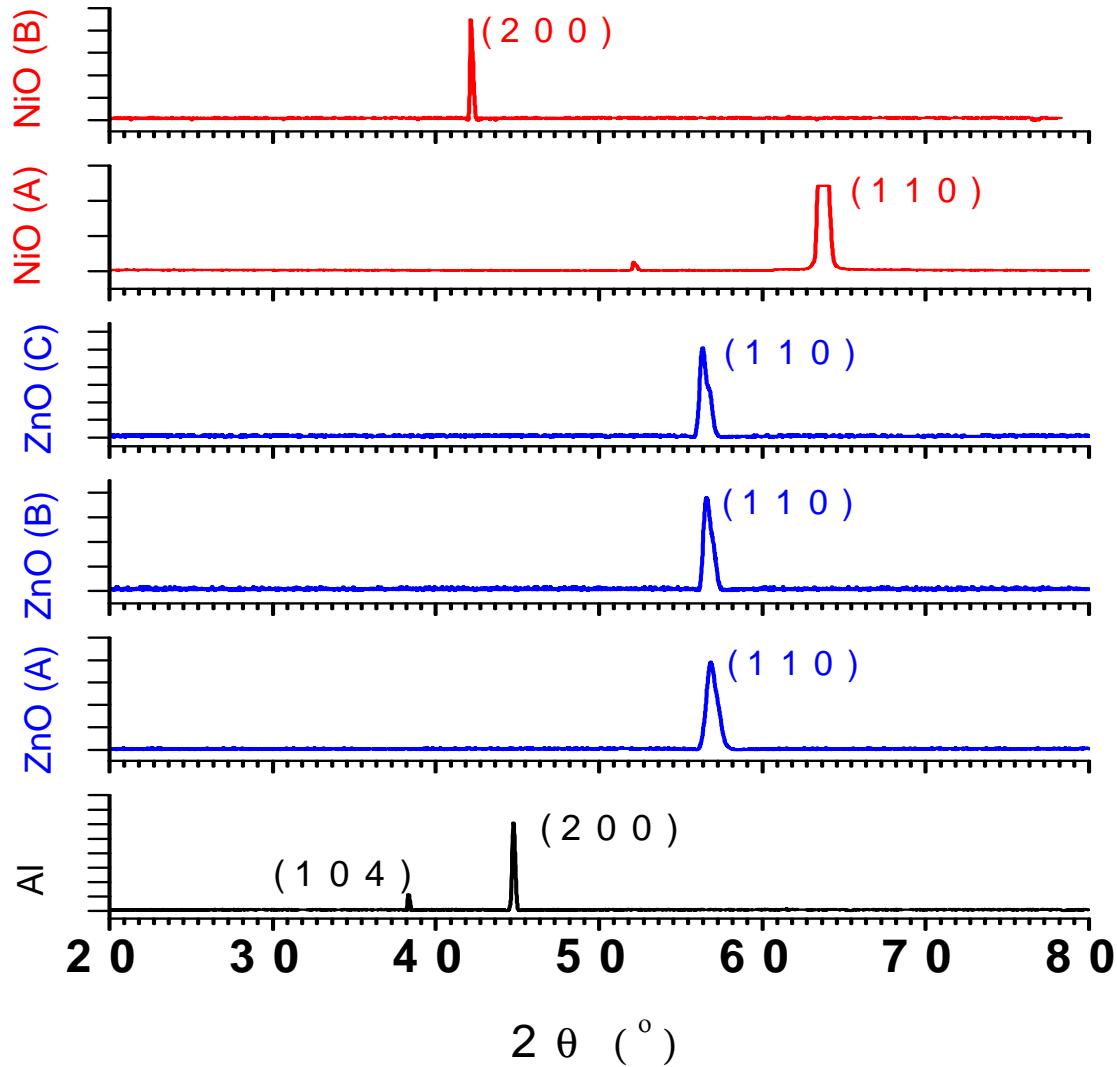
# XRD results: NiO and ZnO



# Experimental results: Comparison



# Experimental results: XRD





## Conclusions

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- There is a possibility to produce low cost selective solar absorbers with sol-gel technique.
  
- Addition of 15 wt.%  $\text{Ac}_2\text{O}$  appeared to solve the problem of cracking in  $\text{SiO}_2$  samples better than 20 wt.% MTES.



## Conclusions

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- New and interesting microstructure of the sol-gel derived samples have been revealed:  
A short chain-like structure of both a silica matrix and carbon nanoparticles is quite evident.
- Homogeneity of the coatings at nano-scale is very encouraging.

# Conclusions

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Sample	$\alpha$	$\varepsilon$	$\alpha/\varepsilon$
C-SiO <sub>2</sub>	0.90	0.31	2.90
C-ZnO	0.89	0.14	6.29
C-NiO	0.93	0.10	8.94

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## Acknowledgement

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